# Metabolism

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## Lecture Outline

- The laws of thermodynamics 热力学定律
- Enzymes 酶
- ATP
- Cellular respiration 细胞呼吸

#### The Energy of Life

- The living cell is a miniature chemical factory where thousands of reactions occur
- Some organisms convert energy to light, as in bioluminescence 生物发光



# Energy

- Kinetic energy 动能 is energy associated with motion
- Heat (thermal energy) 热能 is kinetic energy associated with random movement of atoms or molecules
- Potential energy 势能 is energy that matter possesses because of its location or structure
- Chemical energy 化学能 is potential energy available for release in a chemical reaction



#### The 1<sup>st</sup> and 2<sup>nd</sup> Laws of Thermodynamics

- According to the first law of thermodynamics, the energy of the universe is constant
  - Energy can be transferred and transformed, but it cannot be created or destroyed
- According to the second law of thermodynamics
  - o Every energy transfer or transformation increases the entropy 熵 (disorder) of the universe



(a) First law of thermodynamics



(b) Second law of thermodynamics

# Metabolism 代谢

- Metabolism is the totality of an organism's chemical reactions
- A metabolic pathway 代谢通路 begins with a starting molecule and ends with a product, and each step is catalyzed by a specific enzyme
- Catabolism (分解代谢) releases energy, and anabolism (合成代谢) consumes energy



### Free-Energy Change, $\Delta G$

- ・ Gibbs Free Energy: G. 吉布斯自由能
- △G: G(products产物)-G(reactants反应物)
- An exergonic reaction 放能反应 △G<0</li>
- An endergonic reaction 吸能反应 △G>0



## Exergonic reaction 放能反应

- Biochemical reactions that are energetically favored **DO NOT** happen spontaneously. WHY?
  - Intermediate state 中间态: much higher free energy than reactants
  - E<sub>A</sub>: energy of activation 活 化能



# How Enzymes Speed Up Reactions

- Enzymes **DO NOT** change free energy of reactants or products
- Enzymes **DO NOT** affect
  ΔG
- Enzymes **DO NOT** affect
  equilibrium
- Enzymes catalyze (speed up) reactions by lowering the E<sub>A</sub> barrier



# Enzymes

- The reactant that an enzyme acts on is called **substrate 底物**
- The enzyme binds to its substrate, forming an enzyme-substrate complex 酶-底物复合体

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- The reaction catalyzed by each enzyme is **very specific 高特异性**
- The active site 活性位点 is the region on the enzyme where the substrate binds



## Animation: How Enzymes Work



#### Effects of Temperature and pH

Each enzyme has an optimal temperature and pH in which it can function

 Optimal conditions favor the most active shape for the enzyme molecule



#### Enzymes in the Cell

• Enzymes for cellular respiration are located in mitochondria线粒体



#### ATP (三磷酸腺苷)

- Adenosine triphosphate (ATP): "能量货币"
- ATP stores the potential to react with water, a reaction that releases energy to be used by the cell (ATP->ADP, release energy: 30.5kJ/mol)



## The carbon cycle



## Cellular Respiration 细胞呼吸

- Step1: Glycolysis 糖酵解 (breaks down glucose into two molecules of pyruvate and produce ATP; O<sub>2</sub> not needed)
- Step 2: Pyruvate oxidation 丙酮酸氧化, citric acid cycle柠檬酸 循环 and Oxidative phosphorylation 氧化磷酸化 (O<sub>2</sub> needed)

## Glycolysis 糖酵解

- Glycolysis breaks down glucose 葡萄糖 into two molecules of pyruvate 丙酮酸
- Glycolysis occurs in the cytoplasm 细胞质 and has two major phases
  - Energy investment phase
  - Energy payoff phase
- Glycolysis does NOT require O2



## Pyruvate oxidation 丙酮酸氧化

- In the presence of O<sub>2</sub>, pyruvate 丙酮酸 enters the mitochondrion 线粒体 where the oxidation of glucose is completed
- Before the citric acid cycle can begin, pyruvate must be converted to acetyl Coenzyme A (acetyl CoA,乙酰辅酶A), which links glycolysis to the citric acid cycle 柠檬酸循环



### The Citric Acid Cycle 柠檬酸循环

- The citric acid cycle, also called the Krebs cycle, completes the break down of pyruvate to CO<sub>2</sub>
- The cycle oxidizes organic fuel derived from pyruvate, generating 1 ATP, 3 NADH, and 1 FADH<sub>2</sub> per turn





#### Oxidative phosphorylation 氧化磷酸化

 NADH and FADH<sub>2</sub> transfer electrons to the electron transport chain 电子传 递链, which powers ATP synthesis via oxidative phosphorylation氧化磷酸化



### Electron Transport 电子传递链

- 电子传递链使电子势能转变为质子跨膜运动,
  将质子从线粒体基质泵入线粒体内膜中
- 质子跨膜运动,产生电位差
- 质子通过ATP合成酶回返线粒体基质,伴随 ATP生成,这一过程又称化学渗透
- $2H^++2e^-+1/2O_2=H_2O$
- 电子传递链不直接产生ATP



### Chemiosmosis 化学渗透

- 圆柱状蛋白质复合物嵌入膜内,充当分子马达。
  当质子流经过其内部时可推动马达顺时针转动。
- 转轴与分子马达和催化蛋白质相连,可与分子马达一道转动并激活催化蛋白质。
- 催化蛋白质位于基质一侧,可将无机磷酸根连接 到ADP上生成ATP。
- Oxidation of NADH -> pump 10 H<sup>+</sup> -> 3 ATP
- Oxidation of  $FADH_2 \rightarrow pump 6 H^+ \rightarrow 2 ATP$





### **Cellular Respiration**



# Accounting of ATP production



#### Fermentation: 发酵

 Fermentation: Two common types are alcohol fermentation (植物) and lactic acid fermentation (动物)



#### **Animation: Fermentation Overview**



#### Fermentation vs. Cellular Respiration

- Use glycolysis (net ATP =  $\mathbf{2}$ ) to oxidize glucose
- The processes have different mechanisms for oxidizing NADH:
  - Fermentation: an organic molecule (pyruvate or acetaldehyde) acts as a final electron acceptor
  - o Cellular respiration: electrons are transferred to the electron transport chain
- Fermentation: **2** ATP per glucose molecule
- Cellular respiration: **30** ATP per glucose molecule;



### The Versatility of Catabolism

- Glycolysis accepts a wide range of carbohydrates 碳水化合物
- Proteins must be digested to amino acids; amino groups can feed glycolysis or the citric acid cycle
- Fats are digested to glycerol 甘油 (-> glycolysis) and fatty acids 脂 肪酸 (-> acetyl CoA)
- 1g fat can produce more than twice as much ATP as 1g carbohydrate

